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States of matter and intermolecular forces |

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Chapter Summaries - Chemistry Matter and Change

Chapter 1 Section 2 MATTER and Its Properties Matter - Anything that has MASS and

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VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT- a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1 Section 1

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the statement or answers the question. 1.
Atomic radii cannot be measured ...

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and ChangesMatter–Properties and Changes Solutions Manual Chemistry: Matter and Change

- Chapter 3 35 Section 3.1 Properties of Matter pages 70–75 Problem–Solving Lab 1. Explain why the flow of a compressed gas must be controlled for practical and safe use.

Matter–Properties and

ChangesMatter–Properties and Changes

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Ch 6 Study Guide answers

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Matter and Change • Chapter 5 25 Electrons in
Atoms Reviewing Vocabulary Match the
definition in Column A with the term in
Column B. Column A Column B 1. The set of
frequencies of the electromagnetic waves
emitted by the atoms of an element 2. The
minimum amount of energy that can be lost or
gained by ...

Chapter Assessment

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Chemistry: Chapter 2: Matter and Change.
extensive property. intensive property.
extensive properties (examples) intensive
property (examples) a property that depends
on the amount of matter in a sample. a
property that depends on the type of matter
in a sample not... mass, volume, length,
weight.

chemistry matter and change chapter 2

Flashcards and Study ...

Math Skills Transparency Masters Chemistry:
Matter and Change • Chapter 11 31 number of
moles of known substance C H C H C H O O O CO

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CO CO H O H O H O number of moles of unknown
substance O CO H O C H CO H O C H O H O C H O
CO use mole ratio 5 mol O 2 1 mol C 3 H 8 3
mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol C 3
H 8 1 mol C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different

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properties than their component elements.

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